

2023
B.A./B.Sc.
Fifth Semester
 CORE – 11
CHEMISTRY
Course Code: CHC 5.11
 (Organic Chemistry - IV)

Total Mark: 70
Time: 3 hours

Pass Mark: 28

Answer two questions, taking one from each unit.

UNIT-I

1. (a) Write the synthesis of cytosine and uracil. 6
 (b) Write the structure of purine and pyrimidine base. 4
 (c) What are purine bases present in RNA and pyrimidine bases in DNA and write the structure of ribose sugar? 4
2. (a) Write the structure of the cytosine and uracil. 4
 (b) Write the structure of guanosine triphosphate (GTP) and adenosine triphosphate (ATP) and show the formation and structure of polynucleotide. 6
 (c) Show the hydrogen bonding between 2×2=4
 (i) Thymine and Adenine (ii) Cytosine and Guanine

UNIT-II

3. (a) Write Merrifield solid phase synthesis of protein. 3
 (b) Write any two process for synthesis of α-amino acid. 4
 (c) Write activation of carboxyl group by conversion in to p-nitro phenyl ester and dicyclohexyl carbodiimide (DCC) derivative. 7
4. (a) Write short notes on the following: 2×3=6
 (i) Denaturation of protein (ii) Zwitter ions
 (iii) Isoelectric point
 (b) Write Sanger's and Dansyl chloride methods for N-terminal end analysis for determination of primary structure of peptides. 3+3=6

- (c) Write the classification of amino acids on the basis of necessity with examples. 2

UNIT-III

5. (a) Write the mechanism of enzyme action by taking trypsin as example. 5
(b) Explain the factors affecting enzyme action. 5
(c) Explain specificity and stereospecificity of enzyme action. 4
6. (a) Explain the terms coenzyme and cofactors. Write their role in biological reactions. 5+5=10
(b) Write note on uncompetitive enzyme inhibitor. 4

UNIT-IV

7. (a) A triglyceride has molecular weight 884 and contains three double bonds. Calculate its saponification number and iodine number. 4+4=8
(b) Fats are solid whereas oils are liquid. Explain why. 3
(c) Write the differences between animal and plant fat. 3
8. (a) Explain drying and non drying oils. 3
(b) A triglyceride has molecular weight 928 and contains four double bonds. Calculate its saponification number and Iodine number. 4+4=8
(c) Explain trans fat. 3

UNIT-V

9. (a) Write the classification of pharmaceutical compounds. 4
(b) Write the synthesis of paracetamol and ibuprofen with its uses. 5+5=10
10. (a) Explain the following terms with suitable examples and their uses
(i) Antipyretics (ii) Analgesics 3×4=12
(iii) Antibiotics (iv) Antacids
(b) What are the side effects of antimalarial drugs. 2