#### 2022

### B.A./B.Sc. Fifth Semester CORE – 11 CHEMISTRY Course Code: CHC 5.11

(Organic Chemistry - IV)

Total Mark: 70 Time: 3 hours Pass Mark: 28

Answer five questions, taking one from each unit.

### UNIT-I

| 1. | (a) | Write the structure of adenine and thiamine.                      | 4    |
|----|-----|---|------|
|    | (b) | Write the chemical composition of nucleotides and show the struct | ture |
|    |     | of different types of sugars present in it.                       | 6    |
|    | (c) | Write the structure of adenosine monophosphate and adenosine      |      |
|    |     | diphosphate.  | 4    |
| 2. | (a) | Write the structure of purine, adenine, and guanine.              | 6    |
|    | (b) | Write the synthesis of adenine and guanine.                       | 5    |
|    | (c) | Write the structure of thymidine monophosphate.                   | 3    |

#### UNIT-II

| 3. | (a) | Write Gabriel phthalimide synthesis for preparation of $\alpha$ -amino aci | d. |
|----|-----|--|----|
|    |     |  | 3  |
|    | (b) | Write Schlack-Kumpf and reductive method for C-terminal analysis           | is |
|    |     | of amino acids.  | 6  |
|    | (c) | Write the general reactions involved in protection of amino and            |    |
|    |     | carboxylic group of amino acids.   | 5  |
| 4. | (a) | Discuss the structure of proteins.   | 7  |
|    | (b) | Write the classification of proteins on the basis of chemical              |    |
|    |     | composition.   | 3  |
|    | (c) | Write the problems posed during the synthesis of peptides from             |    |
|    |     | $\alpha$ -amino acid.  | 4  |
|    |     |  |    |

# UNIT-III

| 5. (a) Write the classification of enzymes.                 | 4     |
|---|-------|
| (b) Explain the salient features of active site of enzymes. | 4     |
| (c) Write short notes on the following:                     | 2×3=6 |
| (i) Non-competitive enzyme inhibitor                        |       |
| (ii) Allosteric inhibition                                  |       |
| (iii) Competitive inhibition                                |       |
| 6. (a) Write short notes on the following:                  | 3×2=6 |
| (i) Enzyme activators                                       |       |
| (ii) Isoenzyme  |       |
| (b) Explain the characteristics of enzyme.                  | 4     |
| (c) Write the applications of enzymes.                      | 4     |

# UNIT-IV

| 7. | (a) | A triglyceride has molecular weight 890 and contains four double   |    |
|----|-----|--|----|
|    |     | bonds. Calculate its saponification number and iodine number.  |    |
|    |     | 4+4  | =8 |
|    | (b) | What are oils and fats? How do they differ from each other?  | 3  |
|    | (c) | Explain the term rancidity of fats and oils.   | 3  |
| 8. |     | Explain the term hydrogenation of oils with suitable example.<br>A triglyceride has molecular weight 800 and contains three double | 4  |
|    |     | bonds. Calculate its saponification number and iodine number.  | 7  |
|    | (c) | Write the biological importance of triglycerides.  | 3  |
|    |     |  |    |

## UNIT-V

| 9.  | (a) | Define the term antipyretics and write its application on the ba | asis of | 2  |
|-----|-----|--|---------|----|
|     |     | chemical structure.  | 1+7=    | =8 |
|     | (b) | Write the chemical name, source, uses and structure of vitami    | n C.    | 4  |
|     | (c) | Explain broad and narrow spectrum antibiotics.                   |         | 2  |
| 10. | (a) | Write the synthesis and uses of chloramphenicol.                 | 6+2=    | =8 |
|     | (b) | Write the preparation of chloroquine and its uses.               | 4+2=    | =6 |