

2024
M.Sc.
Fourth Semester
 CORE – 11
CHEMISTRY
Course Code: MCHC 4.11
 (Inorganic Chemistry - IV)

Total Mark: 70
Time: 3 hours

Pass Mark: 28

Answer five questions, taking one from each unit.

UNIT-I

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| 1. | (a) Mention the biological role of iodine in biosystems. | 3 |
| | (b) Discuss the structure and function of biological membranes. | 5 |
| | (c) Discuss the Na ⁺ - K ⁺ pump, its mechanism and functions in biosystems. | 6 |
| 2. | (a) What are ionophores? | 4 |
| | (b) Discuss the crown ether complexes of Na ⁺ and K ⁺ . | 5 |
| | (c) Explain the PS I and PS II mechanisms in photosynthesis. | 5 |

UNIT-II

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| 3. | (a) What are hemeproteins? Draw the structure of heme group and discuss how iron is bonded to different types of ligands in hemeproteins. | 1+2+3 =6 |
| | (b) Discuss the optical spectra of haemoglobin under different PO ₂ levels. | 5 |
| | (c) Write short notes on the synthetic oxygen carrier with reference to Vaska's Ir complex. | 3 |
| 4. | (a) Discuss the magnetic susceptibility of hemeprotein. | 5 |
| | (b) With the help of diagram, explain the EPR spectrum of isolated ferric alpha chains in the hydroxide form and the high field portion of dihistidyl form. | 6 |
| | (c) Discuss the evidence of Fe(III) in myoglobin. | 3 |

UNIT-III

5. (a) Explain the role of carbonic anhydrase in Zn enzymes. 5
(b) Write notes on blue copper proteins. 4
(c) Discuss the importance of iron metallo-enzymes with reference to its storage and transport. 5
6. (a) Write short notes on the following: $3 \times 2 = 6$
(i) Ferrichrome
(ii) Sederophores
(b) Define epoenzyme and coenzyme. $1 + 1 = 2$
(c) Mention two metalloenzymes of each of the following metals: $1 \times 3 = 3$
(i) Fe
(ii) Cu
(iii) Zn
(d) Write notes on cytochrome oxidase. 3

UNIT-IV

7. (a) Explain platinum-DNA binding with diagram. 4
(b) Write notes on the following: $3 \times 2 = 6$
(i) Iron deficiency
(ii) Minamata disease
(c) What are the symptoms of Hg poisoning? 4
8. (a) Discuss the uses of cisplatin in medical field. 4
(b) Discuss arsenic poisoning in rural areas in India. 5
(c) Explain the symptoms and prevention of Be toxicity. 5

UNIT-V

9. (a) Discuss the photosubstitution reactions of Cr (III) complexes. 4
(b) Give the differences between fluorescence and phosphorescence. 4
(c) Write short notes on the following: $3 \times 2 = 6$
(i) Ligand field states
(ii) Thexi states

10. (a) Discuss the photoredox reactions of Cr (III) complexes. 4
- (b) How many major structural types are observed for dinuclear metal clusters containing M-M multiple bonds? Explain edge sharing bio-octahedra. 1+3=4
- (c) Write short notes on the following: 3×2=6
- (i) Photochromism
 - (ii) Charge transfer states
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