

2023
M.Sc.
Second Semester
CORE – 05
CHEMISTRY
Course Code: MCHC 2.11
(Inorganic Chemistry - II)

Total Mark: 70
Time: 3 hours

Pass Mark: 28

Answer five questions, taking one from each unit.

UNIT-I

1. (a) Discuss one method of preparation each of $\text{Co}_2(\text{CO})_8$ and $\text{Fe}_2(\text{CO})_9$. Draw their structures. 1½ × 2 + 2 = 5
(b) Calculate the number of metal-metal bonds in the following and draw their structure. 3 × 2 = 6
 - (i) $\text{Ir}_4(\text{CO})_{12}$
 - (ii) $\text{Ru}_3(\text{CO})_{12}$
- (c) Write a short note on Wade's rule. 3
2. (a) Find out the structure (closo/nido/arathro) of the following: 2 × 4 = 8
 - (i) $\text{Fe}_5\text{C}(\text{CO})_{15}$
 - (ii) $\text{Rh}_6(\text{CO})_{16}$
 - (iii) $[\text{Fe}_4\text{N}(\text{CO})_{12}]^-$
 - (iv) $\text{Ru}_6\text{C}(\text{CO})_{17}$
- (b) Give one method of preparation of transition metal complex with N_2 ligand. 2
- (c) Write the applications of supramolecular chemistry. 4

UNIT-II

3. (a) On the basis of valence bond theory, discuss the cause of lability and inertness of octahedral complexes. 5
- (b) What is an inert ligand? Discuss what type of intermediates and products will be formed if the inert ligand is a pi-acceptor. 1 + 2 + 2 = 5

- (c) Write notes on fluxional coordination compounds with reference to trigonal bipyramidal molecules. 4
4. (a) Define base hydrolysis of octahedral complexes with an example. Mention any two points that S_N2 mechanism alone cannot explain base hydrolysis. 2+2=4
- (b) Give any two evidence to suggest that substitution in square planar complexes proceeds through S_N2 mechanism. 5
- (c) Discuss the fluxional organometallic compounds. 5

UNIT-III

5. (a) How are Fischer carbenes prepared from electron rich olefins? Give reactions. 3
- (b) Discuss bonding in Schrock carbenes. 3
- (c) What are low valent carbyne complexes? Give one method of preparation and discuss its bonding and structure. 2+2+4 = 8
6. (a) What are $C_3R_3^+$ compounds? How are they prepared? Mention any of its reactions. 2+3+2=7
- (b) Give the reactions of Fischer carbenes. 3
- (c) What are vinylidenes? Give an example. 3+1=4

UNIT-IV

7. (a) What do you mean by cyclopentadiene? Give the chemical reaction for the preparation of metallocenes by the action of anhydrous transition metal(II) halide (MX_2) on alkali metal cyclopentadienides. 1+2=3
- (b) Give the synthesis and reaction of cyclopentadienyl metal halides. 2
- (c) Write the physical properties of arene complexes with special reference to $(\eta^6-C_6H_6)_2Cr$. 2
- (d) Write the reactions of the following for ferrocene: $2\frac{1}{2} \times 2 = 5$
- (i) Friedel-Craft's alkylation
- (ii) Mannich condensation amino methylation
- (e) Give the reaction of $Cr(\eta^6-C_6H_6)_2$ with alkyl halides. 2

8. (a) Explain the structure and bonding in ferrocene. 4
- (b) Write the metallation and substitution reactions of $\text{Cr}(\eta^6\text{-C}_6\text{H}_6)_2$.
2+2=4
- (c) Give the Fischer's reducing Friedel Craft's method for the preparation of arene metal complexes. 2
- (d) Give the physical properties of ferrocene. How ferrocene can be prepared in the laboratory? 2+2=4

UNIT-V

9. (a) Write note on hydrozirconation of alkene and alkyne. 4
- (b) Explain with mechanism the synthesis of acetaldehyde from ethene using tetrachloropalladate(II) ion as catalyst. 5
- (c) Briefly explain the vinylation of olefins with mechanism by Heck's reaction. 5
10. (a) Discuss the applications of η^4 -diene iron tricarbonyls in organic synthesis. 5
- (b) Write notes on the following: 3×2=6
- (i) Asymmetric epoxidation
- (ii) Asymmetric hydrogenation
- (c) Explain the synthesis of acetic acid by Monsanto process. 3