2023 B.A./B.Sc. Sixth Semester CORE - 14CHEMISTRY Course Code: CHC 6.21 (Organic Chemistry - V)

Total Mark: 70 Time: 3 hours

(iii)

Answer five questions, taking one from each unit.

# UNIT\_I

- 1. (a) Explain the following terms:
  - (i) Chromophore
  - (iii) Bathochromic shift
  - (b) On the basis of Woodward rules calculate the absorption maximum  $2 \times 3 = 6$

(ii) Auxochrome

 $(\lambda_{max})$  of the following compound:



- 2. (a) Explain types of fundamental vibrations.
  - (b) A compound with molecular weight 130 gave a negative iodoform test. In its IR spectrum the various bands are at 3042, 2941, 2862, 2740, 1722, 1605, 1575, and 1462 cm<sup>-1</sup>. Write the structural formula of the compound. 4

Pass Mark: 28



 $2 \times 3 = 6$ 

2

4

(c) Calculate the  $\lambda_{max}$  of the following compounds:



#### **UNIT-II**

- 3. (a) How many NMR signals would you expect from the following?  $1 \times 4=4$ 
  - (i)  $(CH_3)_3C-NH_2$  (ii)  $CH_3-CH_2-O-CH_2-CH_3$
  - (iii)  $CH_3CH_2CHO$  (iv)  $CH_3-CH_2-O-CH_2-COOH$
  - (b) An organic compound with molecular formula  $C_3H_8O$  gives the following NMR data
    - (i) Three proton triplet  $(9.1\tau)$  (ii) Two proton sextet  $(8.45\tau)$
    - (iii) Two proton triplet  $(6.4\tau)$  (iv) One proton singlet  $(7.7\tau)$ Assign the name of the compound.
  - (c) An organic compound with molecular formula  $C_8H_8O$  gives the following NMR data
    - (i) multiplet 2.72  $\tau$  (5H) (ii) doublet 7.2  $\tau$  (2H)
    - (iii) triplet 0.22  $\tau$ (1H)

Give the structural formula of the compound.

(d) Which type of nuclei show magnetic properties for the purpose of NMR spectroscopy?

4. (a) Predict the structural formula for the compounds with the following formulae showing one NMR signal each.  $1 \times 4=4$ 

(i)  $C_8 H_{18}$  (ii)  $C_2 H_6 O$ (iii)  $C_5 H_{12}$  (iv)  $C_4 H_8$ 

(b) An organic compound with molecular formula  $C_6H_{12}O_2$  gave the following NMR data

- (i) singlet  $1.18\tau$  (6H) (ii) singlet  $2.18\tau$  (3H)
- (iii) singlet 2.68  $\tau$  (2H) (iv) singlet 3.98  $\tau$  (1H)

Give the structural formula of the compound.

 $2 \times 3 = 6$ 

4

4

2

5

- (c) How will you distinguish inter and intra-molecular hydrogen bonding from NMR spectra? 2
- (d) How many NMR signals are expected for the following compounds?  $1 \times 3 = 3$ 
  - (i) CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>2</sub>-Br
    - (ii) CH<sub>2</sub>-CHCl-CO-CH<sub>2</sub>
    - (iii) CH<sub>3</sub>-CHO

#### **UNIT-III**

5.	(a)	How will you convert the following?	5
		(i) Aldopentose to aldohexose	
		(ii) Aldohexose to aldopentose	
	(b)	Write a short note on mutarotation.	3
	(c)	Starting with glucose, how will you prepare	6
		(i) Sorbitol (ii) Gluconic acid	
		(iii) Glucose phenylhydrazone	
6.	(a)	What are carbohydrates? How are they classified?	1+3=4
	(b)	Write the Haworth projection formula for (+) sucrose and	
		(+) maltose.	4
	(c)	Establish the structure of glucose.	6

### **UNIT-IV**

7.	(a)	Write the synthesis of methyl orange. Give its structure in acid at	e in acid and	
		alkali solution. Write its uses. 3+2	+1=6	
	(b)	Outline the synthesis of the following dyes: 2	×3=6	
		(i) Rosaniline (ii) Crystal violet		
	(c)	How are dyes classified? Give an example of each class of dyes	s. 2	
8.	(a)	Write the synthesis of phenolphthalein and give its benzenoid and	d	
		quinonoid structure and write its uses. 3+2	+1=6	
	(b)	Write the structure elucidation and synthesis of alizarin.	6	
	(c)	What are chromophores? Give examples of the chromophores.	2	

## UNIT-V

9. (a) Write the mechanism for anionic and free radical addition polymerization reaction. 3+3=6

- (b) Define thermoplastic and thermosetting polymers with two examples each. 2+2=4
- (c) In a polymer samples 30% molecules have a molecular mass 20,000, 40% have molecular mass 30,000 and the rest have 60,000.
  Calculate weight average and number average molecular mass.

10. (a) Explain the following with examples:	3×3=9
(i) Addition polymerization	
(ii) Condensation polymerization	
(iii) Synthetic rubber	
(b) Write a short note on vulcanization of rubber.	2
(c) Write the synthesis of Buna-S rubber and its uses.	3