2022

M.Sc.

Second Semester

CORE - 06

CHEMISTRY

Course Code: MCHC 2.21 (Organic Chemistry – II)

Total Mark: 70 Pass Mark: 28

Time: 3 hours

Answer five questions, taking one from each unit.

UNIT-I

- (a) Explain homogeneous catalytic hydrogenation with Wilkinson's catalyst.
 (b) Write the reduction mechanism of alkyne with sodium metal and liquid ammonia.
 (c) Explain Birch reduction with mechanism.
 (d) Complete the following reaction
 Me
 - Me | Na CH₂=C-CH=CH₂ liq. NH₃
- 2. (a) Explain heterogenous catalytic hydrogenation of alkene.4 (b) Explain the reduction of benzophenone in presence and absence of proton donor.
 - (c) Write the function of catalyst.
 - (d) Explain the term promoters and poisons.

UNIT-II

- 3. (a) Write concrete mechanism for hydroboration of alkene. 4
 - (b) Explain regioselectivity and stereochemistry for hydroboration of alkene.
 - (c) Write the conversion of boranes to 2° alcohols, 3° alcohols and ketone.

6

4

- 4. (a) Write the mechanism of oxidation of 2° alcohol to ketone by H₂CrO₄
 - (b) Explain Prevost and Woodward hydroxylation of alkenes. 6

4

4

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(c) Write oxidative cleavage of C–C bond by lead tetra-acetate. 4

UNIT-III

- 5. (a) What are carbenes? Give its structure. Write the carbene cyclopropanation reaction.
 - (b) State and explain Wolff rearrangement reaction and its reaction mechanism.
 - (c) Bring out the following transformation involving nitrene intermediate.

(i)
$$\frac{\text{NsNH}_2}{\text{Ph(OAc)}_2}$$
 $\frac{\text{H}}{\text{N}}$ $\frac{\text{Ns}}{\text{H}}$

(ii)
$$N=C=0$$

$$0$$

$$N=C=0$$

$$2\times 3=6$$

(iii)
$$NO_2$$
 O N_3 OH NO_2 O NO_2 OH NO_2 OH NO_2 OH

- 6. (a) State and explain Curtius rearrangement reaction.
 - (b) Give the possible product of the following reactions and its mechanism. $3\times 3=9$

(i)
$$R^1$$
 R^2 HN_3, H^+

(ii)
$$R$$
 $Cu, PhH, Reflux$ R

(iii)
$$\begin{array}{c} O \\ \parallel \\ C \end{array}$$
 SoCl₂, RCHN₂ OH

UNIT-IV

- 7. (a) What are free radicals? Discuss the radical cyclization of 5-hexenyl radicals.
 - (b) Write the product of the following reactions. $2\times2=4$

(i)
$$\underbrace{NBS}_{hv \text{ or heat}}$$

- (c) Benzyne possess electrophilic character and undergoes reaction with nucleophile. Justify.
- (d) What is NBS? Give one application of NBS.

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- 8. (a) Discuss the nucleophilic addition to arynes.
 - (b) What are arynes? Give two methods to generate arynes?
 - (c) Give the reaction mechanism of the following reaction. $4 \times 2 = 8$

(i)
$$N N M Pb(OAc)_4$$
H

(ii) Br
$$EtO$$
 OEt OEt OEt OEt OMe OMe

UNIT-V

9. (a) State and explain the following name reactions.

 $4 \times 2 = 8$

- (i) Henry reaction
- (ii) Zimmerman-Traxler model
- (b) Bring out the following transformation

 $3 \times 2 = 6$

(ii)
$$P \longrightarrow O$$
 OEt O OET O

- 10. (a) State and explain with mechanism the following reactions. $4\times2=8$
 - (i) Prins reaction
 - (ii) Pictet-Sprengler reaction
 - (b) How will you bring about the following transformation? $3\times2=6$

(i)
$$\begin{array}{c|c} CH_2=CH_2 \\ \hline (Pd) \\ \hline \\ (ii) \end{array} \begin{array}{c} CH=CH_2 \\ \hline OH \ Cl \\ \hline \\ 2. \ Et_3N \end{array}$$